

Asymmetric Synthesis of β -Lactams

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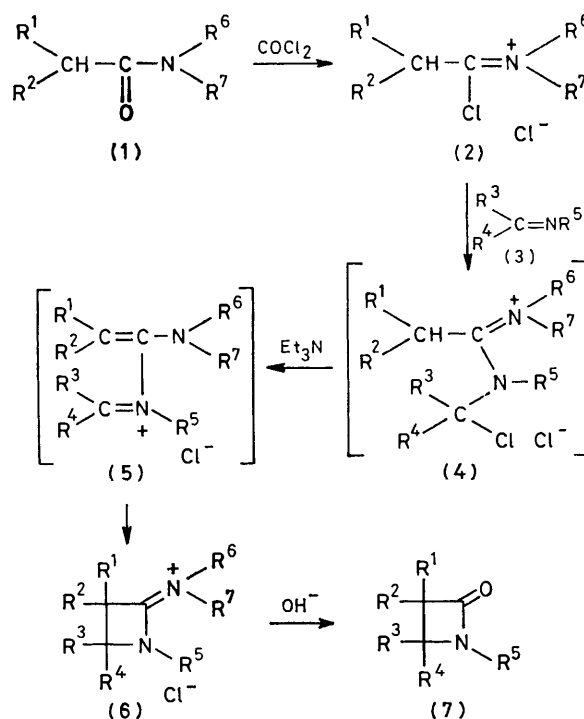
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Summary The reaction of α -chloroiminium chlorides with imines in which one of reagents contains a chiral substituent leads diastereoselectively to substituted β -lactams.

From the variety of synthetic methods leading to β -lactam systems only a few have been adopted for asymmetric syntheses.¹⁻³ The reaction of α -chloroiminium chlorides with imines reported by Ghosez *et al.*^{4,5} seemed as if it would be very useful for the diastereoselective synthesis of substituted β -lactams.

The inducing chiral centre may be present in either or both substrates. Thus R^6 in the amide (1) can be chiral as well as R^5 in the imine (3) (Scheme), and we have examined both possibilities. The reaction of achiral amides (1, $R^1 = R^2$) with chiral imines (3, R^5 chiral) yielded nonequimolar mixtures of two epimeric β -lactams (7) with a new chiral centre at C-4. A similar reaction with the racemic amide (1, $R^1 \neq R^2$), which becomes sp^2 -prochiral in the intermediate (5), however, gave a mixture of the four possible diastereoisomers with new chiral centres at C-3 and C-4.

The second approach, *i.e.* the reaction of chiral amides (1, R^6 chiral) with prochiral imines, seemed to be more interesting. The inducing centre can be removed and regenerated easily by hydrolysis of the salt (6) yielding a mixture of epimeric ($R^1 \neq R^2$) or enantiomeric ($R^1 = R^2$) β -lactams (7). The β -lactams (7) were prepared by the procedure in refs. 4 and 5. The appropriate amide (1) was treated with an excess of $COCl_2$ giving the salt (2), which was treated without separation with 1 mol. equiv. of the imine (3) and then Et_3N . Evaporation led to the iminium salt (6) which was hydrolysed with 1 M aqueous NaOH and, after extraction, purified by chromatography (SiO_2).



SCHEME

The mixtures of diastereoisomers (7a-d, Table) were separated by h.p.l.c. using three $\frac{3}{4}$ in \times 1 ft columns filled with 10μ Lichosorb using hexane containing 15-40% of ethyl acetate as eluant, and a refractive index detector.

TABLE. Yields and diastereoisomeric (enantiomeric) ratios for the β -lactams (7).

	β -Lactam (7)					NR ⁶ R ⁷	Diastereoisomeric (enantiomeric) ratio and $[\alpha]_D^{20}$				Total % yield
	R ¹	R ²	R ³	R ⁴	R ⁵		A	B	C	D	
a	H	H	H	Ph	(S)-(-)-NCHMePh	NMe ₂	1	2.7 ^a			87
b	Me	Me	H	Ph	" "	NMe ₂	-28.1°	+28.5°			85
c	H	Ph	H	Ph	" "	NMe ₂	-95.6°	+90.6°	5	11	64
d	Me	Ph	H	Ph	" "	NMe ₂	+113.0°	-216.7°	1	3	92
e	Me	Me	H	Ph	Me	(+)-N[CH ₂] ₄ CHEt	-27.6°				60
f	Me	Me	H	Ph	Me	(+)-NMeCHMeCH ₂ Ph	+118.0°	6 ^b			30
								-67.3°			

^a A—D are in the order of the R_f value sequence; h.p.l.c. separation and ¹H n.m.r. analysis. ^b Mixtures of enantiomers, the ratio being determined by ¹H n.m.r. spectroscopy with addition of Eu(tfc)₃ shift reagent. $[\alpha]$ values are for the mixture of enantiomers.

The ratio of diastereoisomers was also determined by ¹H n.m.r. integration; compound (7a) (Table), δ (CCl₄) (isomer A): 1.82 (CHMePh, d, J 7.5 Hz) and 4.17 (CHMePh, q, J 7.5 Hz); δ (isomer B): 1.35 (CHMePh, d, J 7.5 Hz) and 4.98 (CHMePh, q, J 7.5 Hz); A:B ratio, 1:2.7; compound (7b) (Table), δ (CCl₄) (isomer A): 3.97 (4-H, s), 1.9 (CHMePh, d, J 7 Hz), 4.2 (CHMePh, q, J 7 Hz), and 0.72 and 1.27 (3-Me₂, 2s); δ (isomer B): 3.92 (4-H, s), 1.52 (CHMePh, d), 4.87 (CHMePh, q, J 7 Hz) and 0.75 and 1.20 (3-Me₂, 2s); A:B ratio 2:1; compound (7e) (mixture of enantiomers) δ (CCl₄) 4.33 (4-H, s), 2.90 (NMe, s) and 0.82 and 0.98 (3-Me₂, 2s); the enantiomeric ratio of 1:9 was determined

from the ¹H n.m.r. spectrum following the addition of Eu(tfc)₃ {tris-[3-(2,2,2-trifluoro-1-hydroxyethylidene)-(+)-camphorato]europium}.

For pure compounds the specific optical rotations are given in the Table; for compounds (7e) and (7f) the rotation is that for the mixture of enantiomers. For compound (7c) the *cis* and *trans* configurations for the A,B and C,D diastereoisomers, respectively, was determined on the basis of the coupling constants (2.5 and 0.0 Hz) for 3- and 4-H.

(Received, 6th August 1980; Com. 871.)

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